

$$C_{\max} = \frac{M}{A * \sqrt{4 * \pi * K * \frac{x}{V}}} * \exp(-k * t)$$

The result leads to Equation (6-7) for  $k = 0$  (no reaction).

2- The standard deviation of the contaminant cloud equals :  $\sigma = \sqrt{2 * K * t}$ .

### *Sudden increase in mass concentration at the origin*

The concentration is initially zero everywhere. At the initial time  $t = 0$ , the concentration is suddenly raised to  $C_0$  at the origin  $x = 0$  and held constant :  $C_m(0, t \geq 0) = C_0$ . The dispersion equation becomes :

$$V * \frac{\partial C_m}{\partial x} = K * \frac{\partial^2 C_m}{\partial x^2} - k * C_m \quad (7-13)$$

with the boundary condition :  $C_m = 0$  at  $x = +\infty$ . A solution of the dispersion equation is :

$$C_m = C_0 * \exp\left(-\frac{k * x}{V} * \left(\frac{2}{\alpha} * (\sqrt{\alpha + 1} - 1)\right)\right) \quad (7-14)$$

where :

$$\alpha = \frac{4 * K * k}{V^2}$$

and  $C_0$  is an integration constant. Neglecting the existence of the initial zone, the initial concentration  $C_0$  is equivalent to the steady release of  $\dot{M}$  units of mass per unit time into a river (discharge  $Q$ ) :

$$C_0 = \frac{\dot{M}}{Q} * \left(\frac{2}{\alpha} * (\sqrt{\alpha + 1} - 1)\right) \quad (7-15)$$

### **Remarks**

1-  $\alpha = 4 * K * k / V^2$  is a dimensionless coefficient of reaction-dispersion. For biochemical oxygen demand,  $\alpha$  is typically of the order :  $\alpha < 0.4$ .

2- In natural rivers and using Equations (5-6) and (6-8), it yields :

$$\alpha = \frac{4 * K * k}{V^2} = 0.0264 * \frac{W^2 * k}{\epsilon_t}$$

where  $W$  is the channel width and  $\epsilon_t$  is the transverse mixing coefficient.

### *Discussion*

The one-dimensional dispersion analysis (Eq. (7-10)) assumes that the contaminant is fully-mixed across the channel : i.e., it does not apply to the initial zone (Fig. 5-1). For a steady flow in rivers, the distance required for cross-section mixing is about (section 5.3) :

$$L \sim 0.4 * V * W^2 / \epsilon_t \quad \text{complete mixing of side discharge}$$

while the characteristic distance of contaminant decay is  $V/k$  (1). The latter exceeds the cross-section mixing distance (i.e.  $x' = 0.4$ ) only if the dimensionless coefficient of reaction-dispersion  $\alpha$  is less than about 0.066.

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<sup>1</sup>Remember that the time required for decay of the contaminant to a factor  $e^{-1}$  is  $1/k$ , during which the fluid is advected to a distance  $V/k$ .